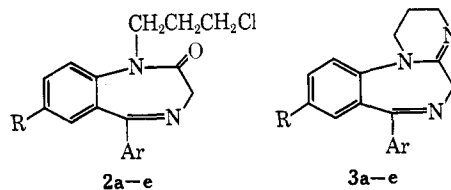


TABLE I
 1-SUBSTITUTED 1,4-BENZODIAZEPINES


Compd	R	Ar	Method	Recrystn solvent ^a	Mp. °C	Yield, %	Formula	Analyses ^b
2a	Cl	C ₆ H ₅	I	A-B	87-90	72.3	C ₁₈ H ₁₆ Cl ₂ N ₂ O	C, H
2b	Cl	<i>o</i> -FC ₆ H ₄	I	A-B	86-89	49.5	C ₁₈ H ₁₅ Cl ₂ FN ₂ O	C, H
2c	Br	2-Pyridyl	I	A-C	103-106	41.9	C ₁₇ H ₁₃ BrClN ₂ O	C, H, N
2d	CF ₃	C ₆ H ₅	I	A-C	118-123	50.4	C ₁₉ H ₁₆ ClF ₃ N ₂ O	C, H, N
2e ^c	H	C ₆ H ₅	I					
3a	Cl	C ₆ H ₅	II	F-C	155-157		C ₁₈ H ₁₆ ClN ₃	C, H, N
3a·HI				D	270-275	49.3	C ₁₈ H ₁₆ ClN ₃ ·HI	C, H
3b	Cl	<i>o</i> -FC ₆ H ₄	II	E-F	161.5-163		C ₁₈ H ₁₅ ClFN ₃	C, H, N
3b·HI				D-A	286-289	64.7	C ₁₈ H ₁₅ ClFN ₃ ·HI	C, H, N
3c	Br	2-Pyridyl	II	F	178-181		C ₁₉ H ₁₅ BrN ₄	C, H, N
3c·HI				D	272-275	34.5	C ₁₇ H ₁₃ BrN ₄ ·III	C, H, N
3d	CF ₃	C ₆ H ₅	II	F	179-181		C ₁₉ H ₁₆ F ₃ N ₃	C, H, N
3d·HI				D	278-282	38.4	C ₁₉ H ₁₆ F ₃ N ₃ ·HI	C, H
3e	H	C ₆ H ₅	II	A	143-145		C ₁₈ H ₁₇ N ₃	C, H, N
3e·HI				D	293-295	52.3	C ₁₈ H ₁₇ N ₃ ·HI	C, H, N

^a A = Et₂O, B = petroleum ether (bp 30-60°), C = hexane, D = EtOH, E = C₆H₆, F = cyclohexane. ^b Acceptable analytical results were obtained for the elements indicated. ^c The isolation of this compound gave a 40.4% yield of an oil which was converted to **3e** without characterization.

Experimental Section⁵

1-(3-Chloropropyl)-1,3-dihydro-2H-1,4-benzodiazepin-2-ones (2a-e). Method I.—The appropriate 1,4-benzodiazepin-2-one (**1a-e**)⁶ was dissolved in a mixture of DMF and THF (1:1) and this solution, under dry N₂, was treated with excess (10-25%) NaNH₂. The mixture was stirred for 2 hr at 50° and was then treated with a threefold excess of 1-bromo-3-chloropropane and was stirred overnight. The reaction mixture was poured into ice water and extracted with CH₂Cl₂ which was dried and evaporated to dryness. The residue was chromatographed on Woelm neutral alumina (Activity 1) on which the product was always less strongly adsorbed than the starting material.

1,2,3,5-Tetrahydropyrimido[1,2-a][1,4]benzodiazepines (3a-e). Method II.—A mixture of the appropriate 1-(3-chloropropyl)-1,3-dihydro-2H-1,4-benzodiazepin-2-one (**2a-e**), 1 equiv of KI, and excess NH₃ in EtOH was heated at 75° for 10 hr with shaking in a sealed container. The reaction mixture was filtered, removing KI, and the filtrate was concentrated *in vacuo* to a residue which was recrystallized from EtOH to give the product as the hydriodide salt. These salts were neutralized in the usual manner to give the crystalline free bases.

8-Chloro-6-(2-fluorophenyl)-1,2-dihydro-4H-imidazo[1,2-a][1,4]benzodiazepine (4).—A solution of 5.0 g (15 mmol) of 1-(2-aminoethyl)-7-chloro-1,3-dihydro-5-(2-fluorophenyl)-2H-1,4-benzodiazepin-2-one⁷ in 100 ml of EtOH was heated for 24 hr at reflux. The solvent was removed *in vacuo* to give a gum which was then triturated with cyclohexane to give 3.9 g (82.5%) of crystals, mp 174-177°. Recrystallizations from CH₂Cl₂-cyclohexane gave pure product as colorless needles, mp 175-177°. *Anal.* (C₁₇H₁₃ClFN₃) C, H, N, Cl, F.

Acknowledgments.—We are indebted to Mr. S. Traiman for the infrared spectrophotometric determinations and to Dr. Al Steyermark and his staff for the microanalyses.

(5) All melting points were determined on a hot stage microscope and are corrected.

(6) Compound **1a**: L. H. Sternbach and E. Reeder, *J. Org. Chem.*, **26**, 4936 (1961); **1b, e**: L. H. Sternbach, R. I. Fryer, W. Metlesics, E. Reeder, G. Sach, G. Saucy and A. Stempel, *ibid.*, **27**, 2788 (1962); **1c**: R. I. Fryer, R. A. Schmidt, and L. H. Sternbach, *J. Pharm. Sci.*, **53**, 264 (1964); **1d**: G. Saucy and L. H. Sternbach, *Helv. Chim. Acta*, **45**, 2226 (1962).

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Some 4H,12H-Pyrano[2,3-a]phenoxazin-4-ones

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Derivatives of phenoxazine²⁻⁵ and γ -pyrone⁶⁻¹¹ are reported to possess marked physiological activity. It was thought desirable to synthesize some 4H,12H-pyrano[2,3-a]phenoxazine-4-ones which will incorporate both phenoxazine and γ -pyrone moieties in its molecule, for biological evaluation.

Experimental Section¹²

8-Amino-7-hydroxy-2-methylchromone was prepared by reducing 8-nitro-7-hydroxy-2-methylchromone¹³ with sodium hy-

(1) Cleveland Clinic, Cleveland, Ohio.

(2) (a) M. L. Crossley, P. F. Dreisbach, C. M. Hofmann, and R. P. Parker, *J. Am. Chem. Soc.*, **74**, 573 (1952); (b) M. L. Crossley, R. J. Turner, C. M. Hofmann, P. F. Dreisbach, and R. P. Parker, *ibid.*, **74**, 578 (1952); (c) M. L. Crossley, C. M. Hofmann, and P. F. Dreisbach, *ibid.*, **74**, 584 (1952); (d) B. Boothroyd and E. R. Clark, *J. Chem. Soc.*, 1499 (1953).

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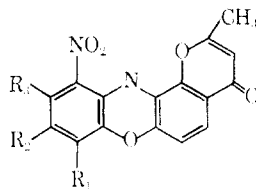
(10) V. V. S. Murti, N. V. S. Rao, and T. R. Seshadri, *Proc. Indian Acad. Sci.*, **25A**, 22 (1947); V. V. S. Murti, L. R. Row, and T. R. Seshadri, *ibid.*, **27A**, 33 (1948); T. R. Seshadri and N. Vishwanadham, *ibid.*, **25A**, 337 (1947).

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(12) Melting points are taken in capillaries and are uncorrected. Where analyses are indicated only by symbols of the elements analytical results obtained for those elements were within $\pm 0.4\%$ of the theoretical values.

(13) C. B. Thanawalla, S. Seshadri, and P. L. Trivedi, *J. Indian Chem. Soc.*, **36**, 674 (1959).

TABLE I
4H,12H-PYRANO[2,3-a]PHENOXAZIN-4-ONES



No.	Derivative of benzene	R ₁	R ₂	R ₃	Yield, %	Mp, °C	Formula ^a
1	1-Chloro-2,4,6-trinitro-	H	NO ₂	H	80	305	C ₁₆ H ₉ N ₅ O ₇
2	1-Chloro-2,4,6-trinitro-5-methyl-	CH ₃	NO ₂	H	70	290	C ₁₇ H ₁₁ N ₅ O ₇
3	1,2,3-Trichloro-4,6-dinitro-	Cl	NO ₂	H	75	295	C ₁₆ H ₃ Cl ₃ N ₅ O ₇
4	1,4-Dichloro-2,6-dinitro-	H	Cl	H	65	275	C ₁₆ H ₃ Cl ₂ N ₅ O ₇
5	1,2,5-Trichloro-4,6-dinitro-	H	NO ₂	Cl	70	260	C ₁₆ H ₃ Cl ₃ N ₅ O ₇
6	1,2-Dichloro-4,6-dinitro-5-methyl-	H	NO ₂	CH ₃	72	250	C ₁₇ H ₁₁ N ₅ O ₇

^a All compounds were analyzed for C, H, N.

drosulfite, following the procedure of Kaufman, *et al.*,¹¹ mp 254-255° (EtOH).

Anal. Calcd for C₁₆H₉N₅O₇: C, 62.82; H, 4.71; N, 7.32. Found: C, 62.80; H, 4.91; N, 7.40.

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General Method for the Preparation of 4H,12H-Pyrano[2,3-a]phenoxazin-4-ones.—To a solution of a halogenonitrobenzene (0.01 mole), 8-amino-7-hydroxy-2-methylchromone (0.01 mole), and EtOAc was added NaOAc (10 ml, 2 N) under stirring. The mixture was refluxed for 2 hr. After cooling, the pyranophenoxazin-4-ones were filtered, washed (H₂O), and crystallized from AcOH. Compounds prepared in this way are listed in Table I.

Book Reviews

Progress in Biochemical Pharmacology. Volumes 2 and 3. Drugs Affecting Lipid Metabolism. Parts I and II. Edited by D. KRITCHEVSKY, R. PAOLETTI, and D. STEINBERG. S. Karger A. G., Basel, Switzerland, 1967. xii + 520 pp and 532 pp, respectively. 17.5 × 24.5 cm. \$29.80 each volume.

From the title of this series, Progress in Biochemical Pharmacology, one would expect its content to constitute a review series. Somewhat surprisingly, this pair of volumes edited by three noted authorities in the field of lipid metabolism is rather a compilation of the papers presented at the Second International Symposium on Drugs Affecting Lipid Metabolism held in Milan, Sept 1965. This is consistent with the contents of Volumes 1 and 4 (in preparation) which cover Symposia on Radioactive Drugs and Atherosclerosis, respectively.

The table of contents enumerates titles of 54 and 56 papers, respectively, and the names of an array of distinguished scientists. These papers are grouped into categories of (Volume 2) Cholesterol and Atherosclerosis, Plasma Triglycerides and Lipoproteins, Drugs Affecting Plasma Lipids, and (Volume 3) Fatty Acids, Prostaglandins, Free Acid Mobilization, Free Fatty Acid and Triglyceride Transport, and Liposoluble Vitamins. Each of these sections is "introduced by a paper presenting recent developments in basic biochemistry as a background for subsequent papers on drug effects." This provides an exceedingly satisfactory format because of the particular excellence of these discussions. The authors of these papers, W. L. Holmes, N. W. DiTullio, G. Schettler, M. F. Oliver, F. Lynen, B. Samuelsson, D. Steinberg, A. Bizzi, S. Garattini, and D. S. Goodman, are to be congratulated for the particular excellence and lucidity of their presentations.

A general point which merits reflection and discussion by research scientists is the whole question of archival-type publication of symposium proceedings. A complete, as opposed to a selective, record of special lectures at a symposium comprising approximately 100 papers can indeed gather a large amount of current information on a specific subject into one compilation. However, these presentations which are often cited later in the

sense of published work are categorized as such without undergoing the usual, critical refereeing process practiced by standard journals. It may be that careful dissemination of invitations to participants and/or subsequent open discussion, carefully recorded in the proceedings, sometimes covers this point. Still, in this instance, over 100 sets of authors required selection. For the minority of papers for which discussion was indeed recorded there was an average of less than one page of commentary.

In many instances, the work either was or will have to be republished with complete experimental data thus leading to a partial or completely duplicate publication or else leave unrecorded many of the details so helpful to later workers. On the whole, it appears that the policy of the Sixth International Congress of Biochemistry or the IUPAC meetings of recording a few unique and rather extensive plenary lectures by outstanding figures is more satisfactory. It provides a rapid and perceptive general review of milestones for other workers in the field while leaving to normal publication channels the vast majority of first class but more standard papers. If an archival record is desired, then the time factor (a year at most) should be stressed and a complete record of all open postpaper discussion should be kept, thereby best approximating actual attendance at the meeting.

In summary, these books present an apparently complete record of an interesting and meritorious symposium held in Sept 1965. By the time this review appears in mid 1968, on the heels of the next scheduled International Symposium on Drugs affecting Lipid Metabolism, some of this work will have been published in standard journals. If one has an intense and continuing interest in this area and/or a library budget which can carry the \$59.60 cost of the set, then its purchase can be recommended. However, one can speculate that (a) a rapidly published, soft-covered, photo-offset set of volumes or (b) a selected presentation of plenary lectures followed by a listing of other speakers and their abstracts, either at a reduced price, would command a significantly greater response than these books are likely to receive.

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